CLASS:	INDIAN SCHOOL MUSCAT	SUBJECT:C
XI	SECOND PERIODIC TEST	HEMISTRY
	SET - C	
QP.NO.	VALUE POINTS	SPLIT UP
		MARKS
1.	c)+8	1
2.	b) (CH <sub>3</sub> ) <sub>3</sub> C <sup>+</sup> ,	1
3.	Redox couple	1
4.	TLC	1
5.	<ul> <li>allows the flow of current by completing the circuit.</li> <li>It maintains electrical neutrality</li> </ul>	1
6.	a) CI	1
	b) 4- lodopent-2-ene	1
7.	$E^{0} \text{ cell } = E_{\text{cathode}} - E_{\text{anode}} = (-0.25) - (-2.87) = 2.62 \text{V}$	1
	Ca/ Ca <sup>2+</sup>    Ni <sup>2+</sup> / Ni	1
8.	i. 1-Hydroxyhexan-2-one	1
	ii. Methoxybutane	1
9.	$MnO_4^- + 4H^+ + 3e^- \rightarrow MnO_2 + 2H_2O$	1
	$Br^- + 3H_2O \rightarrow BrO_3^- + 6H^+ + 6e^-$	1
	$2MnO_4^- + Br^- + H_2O \longrightarrow 2MnO_2 + BrO_3^- + 2OH^-$	1
10.	<ul> <li>a) known mass of an organic compound is heated with fuming nitric acid in the presence of silver nitrate. Carbon and hydrogen present in the compound are oxidized to carbon dioxide and water. The halogen present forms the corresponding silver halide. It is filtered, dried, and weighed. From the mass of AgX, percentage of halogen can be calculated.</li> <li>b) H<sub>2</sub>O, (CH<sub>3</sub>)<sub>3</sub>N -Nucleophiles Cl<sup>+</sup>, BF<sub>3</sub> - Electrophiles</li> </ul>	2  1/2+1/2
11.	i. CH <sub>3</sub> -CH <sub>2</sub> -OH and CH <sub>3</sub> -O-CH <sub>3</sub> ii. Polarisation of sigma bond by the polarization of adjacent sigma bond iii.	1 ½