

### **INDIAN SCHOOL MUSCAT**

# SENIOR SECTION DEPARTMENT OF CHEMISTRY

### **CLASS XII**

## CHAPTER – ALCOHOLS, PHENOLS AND ETHERS WORKSHEET - 5

### 1. Name the following compounds according to the IUPAC system:

vi) 
$$\begin{array}{c} \operatorname{CH_2OH} \\ \operatorname{CH_3} - \operatorname{CH_2} - \operatorname{CH} - \operatorname{CH} - \operatorname{CH} - \operatorname{CH}_3 \\ \operatorname{CH_2Cl} & \operatorname{CH_3} \end{array}$$

ii) 
$$CH_3$$
 $CH=CH-C-OH$ 
 $CH_3$ 

iii) 
$$NO_2$$
  $OC_2H_5$ 

$$\begin{array}{c} \text{iv)} & \text{CH}_2\text{OH} \\ \text{CH}_3 - \text{CH} - \text{CH}_2 - \text{CH} - \text{CH} - \text{CH}_3 \\ \text{CH}_3 & \text{OH} \end{array}$$

 $C_2H_5OCH_2CH_2CH_2OH$ 

#### 2 Account for the following:

- i) C O H bond angle in alcohol is less than tetrahedral angle.
- ii) C O bond length in phenol is shorter than that in methanol.
- iii) C O C bond angle in ether is greater than the tetrahedral angle.
- iv) The boiling points of alcohols and phenols are higher than corresponding alkanes of same molecular mass.
- v) Among the isomeric alcohols the boiling point follows the order  $3^0 < 2^0 < 1^0$ .
- vi) Lower alcohols are soluble in water
- vii) Ethanol is less acidic than methanol.
- viii) The acidic character of the alcohols follows the order  $1^0 > 2^0 > 3^0$
- ix) The reaction of alcohol with acid is carried out in presence of small amount of concentrated H<sub>2</sub>SO<sub>4</sub>.

- x) Reaction of alcohol with acid chloride is carried out in presence of a base pyridine. Effect the following conversions: 3. i) Chloro benzene to phenol Propene to 1-Propanol vi) ii) Benzene sulphonic acid to phenol Propene to 2-propanol vii) Ethanol to isopropyl alcohol Phenol to anisole iii) viii)
  - v) Phenol to p-hydroxy x) Aniline to phenol acetophenone
- 4. Arrange the following on the increasing property given in bracket:
  - a) Pentan-1-ol, butan-1-ol, butan-2-ol, ethanol, propan-1-ol, methanol (Boiling Point)

ix)

Phenol to aspirin

- b) Pentan-1-ol, n-butane, pentanal, Ethoxyethane (Boiling point)
- c) Propan-1-ol, 2, 4, 6 trinitro phenol, 3,5 dinitro phenol, 4-methylphenol (Acidity)
- d) Ter. Butyl alcohol, isobutyl alcohol, n-butyl alcohol (Acidity)
- e) 4-nitro phenol, phenol, 2,4,6-trinitro phenol (Acid strength)
- 5. Write short note on the following:

Phenol to picric acid

iv)

- i) Hydroboration iv) Kolbe's reaction
- ii) Reimer Tiemann Reaction. v) Williamson's synthesis
- iii) Friedel-Craft's reaction
- 6. Write the mechanism of the following:
  Acid catalyzed dehydration of ethanol to diethyl ether.

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